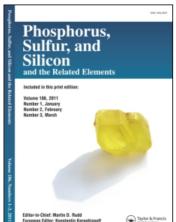
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Phosphorylation as a Method of Stabilization of Thermodynamically Unstable Isomers of A-Amino Ketones

Y. V. Balitskyª; S. E. Pipkoª; T. V. Kolodkaª; A. L. Kovalenkoª a Institute of Organic Chemistry, Ukrainian Academy of Sciences, Kiev, USSR

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PHOSPHORYLATION AS A METHOD OF STABILIZATION OF THERMODYNAMICALLY UNSTABLE ISOMERS OF α -AMINO KETONES

Y.V.BALITSKY, S.E.PIPKO, T.V.KOLODKA, and A.L.KOVALEN-KO

Institute of Organic Chemistry, Ukrainian Academy of Sciences, Murmanskaya Str. 5, Kiev 252660, USSR

 α -hydroxy imines $\underline{2}$ are unstable compounds which irreversibly isomerize into thermodynamically more stable α -amino ketones $\underline{1}$, usually at the moment of formation. We have found the principles of stabilization of both isomeric forms, for example, by regionselective phosphorylation of the ambident substrate 1.

Depending on the structure of the phosphorylating reagent and the reaction conditions (kinetic or thermodynamical control) the interaction leads to the products of N- or O-phosphorylation. As a rule, phosphorotropic isomers $\underline{3}$, $\underline{4}$ are stable enough to be isolated and identified. In contrast to prototropic isomer $\underline{1}$, N-phosphorylated amino ketones $\underline{3}$ on heating at $120\text{-}160^{\circ}\text{C}$ transform into thermodynamically more stable O-derivatives by $[1,4]-(\text{N}\rightarrow\text{O})$ -migration of the phosphorus-containing group. The thermal isomerization is accelerated in the presence of acids. The mild hydrolysis of both O- and N-phosphorylated compounds $\underline{3}$, $\underline{4}$ leads to the α -amino ketone $\underline{1}$ in accordance with the equilibrium of prototropic isomers, completely shifted to the side of N-H form $\underline{1}$.